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1. IDENTIFICATION OF THE SUBSTANCE / MIXTURE AND OF THE COMPANY / UNDERTAKING

Product identifier

Product Name Zephex[™] 134a

Hazardous Ingredient(s)/Substance identity	REACH Registration No.
1,1,1,2-tetrafluoroethane (HFC 134a)	01-2119459374-33-0

Use Subject to Member State regulations, applicable uses are:

medical propellant

Manufacturer Koura Global, (formally Mexichem UK Limited)

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2. HAZARDS IDENTIFICATION

Low acute toxicity. High exposures may cause an abnormal heart rhythm and prove suddenly fatal.

Very high atmospheric concentrations may cause anaesthetic effects and asphyxiation.

Liquid splashes or spray may cause freeze burns to skin and eyes.

Classification of the substance or mixture

Regulation (EC) No. 1272/2008 (CLP)

Gases under pressure - Liquefied gas

Label elements

According to Regulation (EC) No. 1272/2008 (CLP)

Hazard Pictogram(s)



Signal Word(s) Warning

Hazard Statement(s) H280: Contains gas under pressure; may explode if heated.

Precautionary Statement(s) P410+P403: Protect from sunlight. Store in a well-ventilated place.



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3. COMPOSITION / INFORMATION ON INGREDIENTS

Alternative names

HFA 134a, 1,1,1,2-tetrafluoroethane (HFC 134a)

Hazardous Ingredient(s)

Hazardous Ingredient(s)	%(w/w)	CAS No.	EC No.	Hazard symbol(s) and hazard statement(s)
1,1,1,2-tetrafluoroethane (HFC 134a)	100	000811-97-2	212-377-0	GHS04; H280

4. FIRST AID MEASURES



The first aid advice given for skin contact, eye contact, and ingestion is applicable following exposures to the liquid or spray. See also section 11.

Description of first aid measures

Inhalation Remove patient from exposure, keep warm and at rest. Administer oxygen

if necessary. Apply artificial respiration if breathing has ceased or shows signs of failing. In the event of cardiac arrest apply external cardiac massage.

Obtain immediate medical attention.

Skin Contact Thaw affected areas with water. Remove contaminated clothing.

Caution: clothing may adhere to the skin in the case of freeze burns. After contact with skin, wash immediately with plenty of warm water.

If irritation or blistering occur obtain medical attention.

Eye Contact Immediately irrigate with eyewash solution or clean water, holding the

eyelids apart, for at least 10 minutes. Obtain immediate medical attention.

IngestionUnlikely route of exposure. Do not induce vomiting. Provided the patient is

conscious, wash out mouth with water and give 200-300 ml (half a pint) of

water to drink. Obtain immediate medical attention.

Further Medical Treatment Symptomatic treatment and supportive therapy as indicated. Adrenaline

and similar sympathomimetic drugs should be avoided following exposure as cardiac arrhythmia may result with possible subsequent cardiac arrest.

Most important symptoms and effects,

both acute and delayed

High exposures may cause an abnormal heart rhythm and prove suddenly fatal. Very high atmospheric concentrations may cause anaesthetic effects

and asphyxiation.

Indication of any immediate medical attention and special treatment needed

Remove patient from exposure, keep warm and at rest. Administer oxygen if necessary. Apply artificial respiration if breathing has ceased or shows signs of failing. In the event of cardiac arrest apply external cardiac massage.

Obtain immediate medical attention.

5. FIREFIGHTING MEASURES

General HFC 134a is not flammable in air under ambient conditions of temperature

and pressure. Certain mixtures of HFC 134a and air when under pressure may be flammable. Mixtures of HFC 134a and air under pressure should be avoided. Certain mixtures of HFCs and chlorine may be flammable or

reactive under certain conditions.

Extinguishing media As appropriate for surrounding fire.

Keep fire exposed containers cool by spraying with water.

Special hazards arising from

the substance or mixtureThermal decomposition will evolve very toxic and corrosive vapours.

(hydrogen fluoride) Containers may burst if overheated.

Advice for firefighters A self contained breathing apparatus and full protective clothing must

be worn in fire conditions. See Also Section 8



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6. ACCIDENTAL RELEASE MEASURES

Personal precautions, protective equipment

and emergency procedures

Environmental precautions

Ensure suitable personal protection (including respiratory protection)

during removal of spillages. See Also Section 8

Prevent liquid from entering drains, sewers, basements and workpits since

the vapour may create a suffocating atmosphere.

Methods and materials for containment and cleaning up

Provided it is safe to do so, isolate the source of the leak. Allow small spillages to evaporate provided there is adequate ventilation.

Large spillages: Ventilate area. Contain spillages with sand, earth or any suitable adsorbent material. Prevent liquid from entering drains, sewers, basements and workpits since the vapour may create a suffocating

atmosphere.

Reference to other sections

8, 13

7. HANDLING AND STORAGE

Precautions for safe handling

Avoid inhalation of high concentrations of vapours. Atmospheric levels should be controlled in compliance with the occupational exposure limit. Atmospheric concentrations well below the occupational exposure limit can be achieved by good occupational hygiene practice. The vapour is heavier than air, high concentrations may be produced at low levels where general ventilation is poor, in such cases provide adequate ventilation or wear suitable respiratory protective equipment with positive air supply.

Avoid contact with naked flames and hot surfaces as corrosive and very toxic decomposition products can be formed.

Avoid contact between the liquid and skin and eyes.

Avoid venting to atmosphere.

The fluorinated greenhouse gas HFA 134a may be supplied in returnable containers (drums/cylinders). The container contains fluorinated greenhouse gases covered by the Kyoto Protocol. The fluorinated greenhouse gases in containers may not be vented to the atmosphere. Regulation (EU) No. 517/2014 of the European Parliament and the Council

on certain fluorinated greenhouse gases.

Process HazardsThe transfer of liquid HFC134a between containers and to and from processing equipment can result in static generation. Ensure adequate

earthing.

Care must be taken to mitigate the risk of developing high pressures in systems caused by a temperature rise when liquid is trapped between closed valves or in cases where containers have been overfilled.

Conditions for safe storage, including any incompatibilities

Keep in a well ventilated place away from fire risk and avoid sources of heat

such as electric or steam radiators.

Avoid storing near to the intake of air conditioning units,

boiler units and open drains.

Specific use Subject to Member State regulations, applicable uses are:

medical propellant

8. EXPOSURE CONTROLS / PERSONAL PROTECTION

Exposure controls

Occupational Exposure Limits



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Occupational Exposure Limits	CAS No.	LTEL (8 hr TWA ppm)	LTEL 8 hr TWA mg/m³	STEL (ppm)	STEL mg/m³	Note
1,1,1,2-Tetrafluoroethane (HFC 134a)	000811-97-2	1000	4240	-	-	WEL

Appropriate engineering controls Provide adequate ventilation. Atmospheric levels should be controlled in-

compliance with the proposed occupational exposure limit.

Personal protection equipment Wear suitable protective clothing, gloves and eye/face protection.

Respirators

In cases of insufficient ventilation, where exposure to high concentrations of vapour is possible, suitable respiratory protective equipment with

positive air supply should be used.

Eye Protection

Wear protective eyewear (goggles, face shield, or safety glasses).

Gloves

Wear thermal insulating gloves when handling liquefied gases.

9. PHYSICAL AND CHEMICAL PROPERTIES

Form liquefied gas
Colour. colourless
Odour slight ethereal
Solubility (Water) slightly soluble

Solubility (Other) soluble in: alcohols , chlorinated solvents , esters , polyethylene glycol

Boiling Point (° C) -26.2 Melting Point (° C) -101

Vapour Density (Air=1) 3.66 at normal boiling point

Vapour Pressure (mm Hg) 4270 at 20° C Specific Gravity 1.22 at 20° C

10. STABILITY AND REACTIVITY

Reactivity See Section: Possibility of hazardous reactions

Chemical stability Stable under normal conditions.

Possibility of hazardous reactionsCertain mixtures of HFCs and chlorine may be flammable or reactive

under certain conditions. Incompatible materials: finely divided metals,

magnesium and alloys containing more than 2% magnesium.

Can react violently if in contact with alkali metals and alkaline earth metals -

sodium, potassium, barium.

Conditions to avoid Avoid high temperatures.

Incompatible materials Finely divided metals, alkali metals (sodium, potassium), alkaline earth

metals (barium, magnesium), alloys containing more than 2% magnesium

Hazardous decomposition products Hydrogen fluoride by thermal decomposition and hydrolysis.



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11. TOXICOLOGICAL INFORMATION

Acute toxicity / Ingestion Highly unlikely - but should this occur freeze burns will result.

Inhalation / Acute toxicity LC50 (rat) (4 hrs) = 500000 ppm (2080000 mg/m³)

High exposures may cause an abnormal heart rhythm and prove suddenly fatal. Very high atmospheric concentrations may cause anaesthetic effects

and asphyxiation.

Acute toxicity / Skin Contact Unlikely to be hazardous by skin absorption.

Skin corrosion/irritationLiquid splashes or spray may cause freeze burns.Serious eye damage/irritationLiquid splashes or spray may cause freeze burns.

Respiratory irritation Non-irritant

Sensitisation It is not a skin sensitiser.

Repeated dose toxicity

An inhalation study in animals has shown that repeated exposures produce

no significant effects. (50000ppm in rats)

Mutagenicity No evidence of mutagenic effects.

Carcinogenicity A lifetime inhalation study in rats has shown that exposure to 50000ppm

resulted in benign tumours of the testis. The increased tumour incidence was observed only after prolonged exposure to high levels, and is considered not to be of relevance to humans occupationally exposed to HFC 134a at

or below the occupational exposure limit.

Reproductive toxicityNo evidence of reproductive effects. Studies in animals have shown that

repeated exposures produce no teratogenic effects.

Specific target organ toxicity

(single exposure)

Not classified

Specific target organ toxicity

(repeated exposure)

Not classified

Aspiration hazard Not applicable

12. ECOLOGICAL INFORMATION

Toxicity Low toxicity to aquatic organisms.

LC50 (Rainbow trout) (96 hour) = 450 mg/l EC50 (Daphnia magna) (48 hour) = 980 mg/l

Environmental Fate and Distribution High tonnage material produced in wholly contained systems.

High tonnage material used in open systems. Gas.

Persistence and Degradation Decomposed comparatively rapidly in the lower atmosphere (troposphere).

Atmospheric lifetime is 14 years. Products of decomposition will be highly dispersed and hence will have a very low concentration. Does not influence photochemical smog (i.e. is not a VOC under the terms of the UNECE

agreement).

Does not deplete ozone.

Has a Global Warming Potential (GWP) of 1430 (relative to a value of 1 f or carbon dioxide at 100 years) according to Annex I of Regulation (EU)

No. 517/2014 on certain fluorinated greenhouse gases.

Values in Annex I are taken from the fourth assessment report (AR4)

of the Intergovernmental Panel on Climate Change.

United Nations Framework Convention on Climate Change (UNFCCC)

reporting GWP is 1300.



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Bioaccumulative potentialThe product has no potential for bioaccumulation.

Mobility in soil Not applicable

Results of PBT and vPvB assessmentNot classified as PBT or vPvB.

Other adverse effects None known.

Effect on Effluent TreatmentDischarges of the product will enter the atmosphere and will not result in

long term aqueous contamination.

13. DISPOSAL CONSIDERATIONS

Waste treatment methods Best to recover and recycle. If this is not possible, destruction is to be in an

approved facility which is equipped to absorb and neutralise acid gases and

other toxic processing products.

Regulatory Information Disposal should be in accordance with local, state or national legislation.

14. TRANSPORT INFORMATION

Hazard label(s)



Road/Rail

UN No. 3159 ADR/RID Class 2.2

ADR/RID Proper Shipping Name 1,1,1,2-TETRAFLUOROETHANE (REFRIGERANT GAS R 134a)

SEA

IMDG Class 2.2

Marine Pollutant Not classified as a Marine Pollutant

AIR

ICAO/IATA Class 2.2

15. REGULATORY INFORMATION

European Regulations

EC Classification According to Regulation (EC) No. 1272/2008 (CLP)

Gases under pressure - Liquefied gas.

Special Restrictions: The fluorinated greenhouse gas HFA 134a may be supplied in returnable

containers (drums/cylinders). The container contains fluorinated greenhouse gases covered by the Kyoto Protocol. The fluorinated greenhouse gases in containers may not be vented to the atmosphere.

Regulation (EU) No. 517/2014 of the European Parliament and the Council

on certain fluorinated greenhouse gases.



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16. OTHER INFORMATION

This data sheet was prepared in accordance with Regulation (EC) No. 1907/2006.

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Glossary

WEL: Workplace Exposure Limit (UK HSE EH40)

COM: The company aims to control exposure in its workplace to this limit

TLV: The company aims to control exposure in its workplace to the ACGIH limit

TLV-C: The company aims to control exposure in its workplace to the ACGIH Ceiling limit

MAK: The company aims to control exposure in its workplace to the German limit

Sk: Can be absorbed through skin

Sen: Capable of causing respiratory sensitisation

Bmgv: Biological monitoring guidance value (UK HSE EH40)

Hazard Statement(s)

H280: Contains gas under pressure; may explode if heated.

The following sections contain revisions or new statements: 1,2,4,5,6,7,8,10,11,12,13,15



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