

#### **SECTION 1 - IDENTIFICATION**

Product identifier used on the

KLEA®134a

label:

Other means of identification: Fluorocarbon 134a, R-134a, HFC-134a, HFA-134a

Recommended use of the

chemical and restrictions on use: Refrigerant, foam blowing agent, aerosol propellant

Name, address, and telephone number of the chemical manufacturer, importer, or other responsible

party:

<u>United States, Mexico & South America</u> <u>Canada</u>

 Mexichem Fluor Inc.
 Mexichem Fluor Canada Inc.

 4990B ICI Rd. / P.O. Box 30
 5000 Yonge Street, Suite 1901

 St. Gabriel, LA 70776
 Toronto, Ontario, M2N 7E9

 800-424-5532 (US)
 800-275-5532 Ext. 384 or 383

(81) 8156-1711 or 1712 (Mexico)

**Emergency telephone numbers:** 

Medical: 800-298-9164 or 303-389-1418

Transportation: In US, Canada, or South America, call Chemtrec @ 800-424-9300

or 703-527-3887 (call collect)

In Mexico, call SETIQ @ 01-800-00-214-00 (call free from any

place in Mexico) or 01-55-59-15-88 (in Mexico City)

**SECTION 2 – HAZARDS IDENTIFICATION** 

Classification of the chemical: Gases Under Pressure - Liquefied Gas

Simple Asphyxiant

Signal word: Warning

Hazard statement(s): Contains gas under pressure; may explode if heated.

May displace oxygen and cause rapid suffocation.

Precautionary statement(s): Protect from sunlight.

Store in a well-ventilated place.

Pictogram(s):

 $\Diamond$ 

Hazards not otherwise classified: May cause frostbite.

Exposure to high concentrations may cause an abnormal heart rhythm which can be fatal. Very high atmospheric concentrations may cause anesthetic effects such as dizziness, drowsiness.

headaches, and unconsciousness.





#### **SECTION 3 - COMPOSITION/INFORMATION ON INGREDIENTS**

Chemical Name, Common Name, and Synonyms CAS # Concentration

1.1.1.2-tetrafluoroethane

(Klea®134a, Fluorocarbon 134a, R-134a, HFC-134a, HFA-134a)

811-97-2 100%

**SECTION 4 - FIRST AID MEASURES** 

Skin: Immediately wash with plenty of warm water (do not rub). Thaw

affected area with water. Remove contaminated clothing. Caution: clothing may adhere to the skin in case of freeze burns. If symptoms

(irritation or blistering) develop, get medical attention.

Eyes: Immediately flush with plenty of water. After initial flushing, remove

any contact lenses and continue flushing for at least 15 minutes. Hold eyelids open during flushing. Have eyes examined and treated by

medical personnel.

Inhalation: Move victim to fresh air. Keep warm and at rest. If breathing is

labored, give oxygen. If only breathing has stopped, give artificial respiration with a pocket mask equipped with a one-way valve to prevent exposure to product or body fluids. If breathing has stopped AND there is no pulse, give cardiopulmonary resuscitation (CPR). Get

immediate medical attention.

Ingestion: If swallowed, seek medical attention. Do not induce vomiting unless

directed to do so by medical personnel. In case of frostbite, immediately rinse lips and mouth with tepid water for at least 15

minutes. Obtain medical attention promptly.

Note to physician: Provide symptomatic and supportive therapy, as indicated.

Administration of epinephrine or similar sympathomimetic drugs should be with special caution and only in situations of emergency life

support as cardiac arrhythmia may result.

**SECTION 5 - FIRE-FIGHTING MEASURES** 

Fire and explosion hazards: HFC-134a is not flammable in air under ambient conditions of

temperature and pressure. Under conditions of high temperature and

pressure, certain HFC-134a/air mixtures were shown to be flammable. Certain mixtures of HFC-134a and chlorine may be

flammable under some conditions.

Containers may burst under intense heat. Ruptured cylinders may

rocket or fragment. Heavy vapor may suffocate.

Specific hazards arising from the

chemical:

During a fire the product can form toxic and corrosive gases such as

hydrogen fluoride.

Fire-fighting procedures: Move containers from fire area, if it can be done without risk. Fight

fire from a protected location to shield personnel from venting or

ruptured containers.



Suitable extinguishing media: As appropriate for surrounding materials/equipment.

Water spray should be used to cool containers.

Unsuitable extinguishing media: None known

Special protective equipment and precautions for fire-fighters:

Use self-contained breathing apparatus with a full-face piece and

special protective clothing.

Sensitivity to mechanical impact: Not applicable

Sensitivity to static discharge: Not expected to be sensitive to static discharge.

## **SECTION 6 - ACCIDENTAL RELEASE MEASURES**

Personal precautions, protective equipment, and emergency procedures:

This product is a liquefied gas, which exits the container at temperatures capable of causing freeze burns (frostbite).

Precautions should take into account the severity of the leak or spill. Move unprotected personnel upwind of leaking container. Ventilate the spill area. Use recommended personal protection and shut off the leak, if without risk. If possible, elevate leak position to highest point of container (should leak gas, not liquid). Water should never be put

on leak nor should cylinder be immersed.

Methods and materials for containment and cleaning up:

If possible, dike and contain spillage. Prevent liquid from entering sewers, sumps, or pit areas since vapor is heavier than air and can create a suffocating atmosphere. Capture material for recycle or

destruction if suitable equipment is available.

Notify applicable government authority if release is reportable or

could adversely affect the environment.

#### **SECTION 7 - HANDLING AND STORAGE**

Precautions for safe handling: Wear appropriate personal protective equipment. A safety shower

and eyewash station should be nearby and ready for use. This product is a liquefied gas, which exits the container at temperatures capable of causing freeze burns (frostbite). Ensure personnel are trained in handling and storing cylinders. Secure containers at all times. Keep containers closed when not in use. Ensure there is adequate ventilation or use proper respiratory protection in poorly ventilated or confined areas. Avoid causing and inhaling high concentrations of vapor. Atmospheric levels should be controlled to below the occupational exposure limit and kept as low as practicable.

Prevent liquid or vapor from entering sumps or sewers since vapor is

heavier than air and may form suffocating atmospheres.

Do not put mixtures of HFC-134a with air or oxygen under pressure; do not use such mixtures for leak or pressure testing.

Do not heat containers.

Liquid transfers between containers may generate static electricity.

Ensure adequate grounding.

Avoid trapping liquid between closed valves or overfilling containers as high pressures can develop with an increase in temperature. Avoid HFC-134a contact with flames or very hot surfaces.



Conditions for safe storage, including any incompatibilities:

Keep containers tightly closed, in a cool, well-ventilated place. Store

at temperature not exceeding 125°F (52°C.).

Keep containers dry.

Keep away from open flames, hot surfaces, welding operations, and

other heat sources.

Keep away from finely divided metals such as aluminum, zinc, magnesium, and alloys containing more than 2% magnesium. Can react violently if in contact with alkali metals and alkaline earth metals

such as sodium, potassium, or barium.

## **SECTION 8 - EXPOSURE CONTROLS/PERSONAL PROTECTION**

OSHA Permissible Exposure

Limit (PEL):

Not established

American Conference of Governmental Industrial Hygienists (ACGIH) Threshold

Limit Value (TLV):

Not established

American Industrial Hygiene Association (AIHA) Workplace Environmental Exposure Level

(WEEL):

1000 ppm 8-hour TWA

Mexichem Fluor Guideline: 1000 ppm 8-hour TWA

Appropriate engineering controls: Use ventilation to maintain safe levels. Where appropriate

engineering controls are not in place or are inadequate, wear suitable

respiratory equipment.

Eye Protection: Use chemical safety goggles or safety glasses and a face shield

when there is potential for eye contact.

Skin Protection: Take all precautions to prevent skin contact. Use gloves and

protective clothing made of material that has been found by user to be impervious under conditions of use to prevent the skin from becoming frozen from contact with liquid. User should verify impermeability under normal conditions of use prior to general use. Additional protection such as an apron, arm covers, or full body suit

may be needed depending on conditions of use.

Respiratory Protection: Not normally needed if controls are adequate. If needed, use

NIOSH/MSHA approved respirator for organic vapors. For high concentrations and oxygen-deficient atmospheres, use positive

pressure air-supplied respirator.



## **SECTION 9 - PHYSICAL AND CHEMICAL PROPERTIES**

Appearance: Clear, colorless liquefied gas

Odor: Odorless Odor threshold: Not available pH: Not applicable Melting point/freezing point: -108°C (-162.4°F) Boiling point: -26.2°C (-15.2°F) Flash point: Does not flash Evaporation rate: Not available Flammability (solid, gas): Not available

flammability/explosive limits:

Vapor pressure: 4,270 mm Hg at 20°C

Vapor density: 3.6 (air = 1)

Specific gravity (relative density): 1.22 at 20°C

Solubility(ies): Very low in water

Partition coefficient: n-

Auto-ignition temperature:

octanol/water:

Upper/lower

> 743°C (> 1369°F)

1.06

Not applicable

Decomposition temperature: Not available

#### **SECTION 10 - STABILITY AND REACTIVITY**

Reactivity: Reacts with finely divided metals such as aluminum, zinc,

magnesium, and alloys containing more than 2% magnesium. Can react violently if in contact with alkali metals and alkaline earth metals

such as sodium, potassium, or barium.

Chemical stability: Stable under normal conditions

Possibility of hazardous

reactions:

Hazardous polymerization will not occur.

Conditions to avoid: Keep away from heat, sparks, and flame. Avoid high temperatures.

Incompatible materials: Finely divided metals such as aluminum, zinc, magnesium, and alloys

containing more than 2% magnesium. Alkali metals and alkaline earth

metals such as sodium, potassium, or barium.

Hazardous decomposition

products:

Hydrogen fluoride by thermal decomposition and hydrolysis. Oxides of carbon and fluoride may be produced by thermal decomposition.



#### **SECTION 11 - TOXICOLOGICAL INFORMATION**

Information on the likely routes of Inhalation, eye, and skin contact exposure:

Symptoms related to the physical, chemical and toxicological characteristics: Delayed and immediate effects and also chronic effects from short- and long-term exposure:

Inhalation: Vapor is heavier than air. May displace oxygen and cause rapid suffocation. Exposure to high concentrations may cause an abnormal heart rhythm (arrhythmia) under stressful conditions which can be fatal. Very high atmospheric concentrations may cause anesthetic effects such as dizziness, drowsiness, headaches, and unconsciousness.

Ingestion: Liquid will cause freeze burns.

Eye contact: Liquid splashes or spray may cause freeze burns. **Skin contact:** Liquid splashes or spray may cause freeze burns.

Other effects: None anticipated.

Numerical measures of toxicity: LC50: 4 hr. (rat) = 567,000 ppm

LD50: Not applicable

Animal test data: Acute inhalation exposures at very high concentrations of HFC-134a

> have shown central nervous system depression in laboratory animals. Cardiac arrhythmias were seen in dogs exposed to 80,000 ppm HFC-134a for 5 minutes, when followed by an injection of epinephrine. This phenomenon is referred to as cardiac sensitization and is an

increased sensitivity of the heart to epinephrine.

Liquefied material was a slight skin irritant to rats, possibly due to local freezing. Vaporized material is non-irritating. It is not a skin

sensitizer.

No toxicity was seen in rats exposed by inhalation for 6 hours/day, 5 days/week for 13 weeks to concentrations up to 50,000 ppm HFC-

134a.

HFC-134a was not genotoxic when tested in a variety of in vitro and

in vivo tests.

In a two-year carcinogenicity study, there was a slight increase in the incidence of benign testicular tumors in male rats exposed to 50,000 ppm HFC-134a. No increased tumors were seen in female rats or in

male and female mice.

Not a reproductive or developmental toxicant.

Carcinogenicity: Not classified as carcinogenic by NTP, IARC, ACGIH, or OSHA.

Teratogenicity, mutagenicity, other reproductive effects:

None known. For further information see animal test data above.

Toxicologically synergistic

products:

None known. Note that administration of epinephrine or similar sympathomimetic drugs following exposure may result in cardiac

arrhythmia.



#### **SECTION 12 - ECOLOGICAL INFORMATION**

Ecotoxicity: Daphnia 48 hour EC50: 980 mg/l

Rainbow trout 96 hour LC50: 450 mg/l

Persistence and degradability: This product is highly volatile and has low water solubility. It will

rapidly evaporate from water. Decomposes comparatively rapidly in the lower atmosphere (troposphere). Atmospheric lifetime is 14 years. Products of decomposition will be highly dispersed and hence will have a very low concentration. Practically non-biodegradable.

Bioaccumulative potential: Low given its Log  $K_{ow} = 1.06$ 

Mobility in soil: Expected to be mobile in soil.

Other adverse effects: Not a significant contributor to photochemical smog and is not

considered to be a VOC. Is not considered an ozone-depleting

chemical.

#### **SECTION 13 - DISPOSAL CONSIDERATIONS**

Disposal Method: Discarded product is not a hazardous waste under RCRA, 40 CFR

261. However, HFC-134a should be recycled or reclaimed whenever

possible.

Container Disposal: For disposable (DOT 39) cylinders only. Do not distribute, make

available, furnish, or reuse container when emptied of the original product. Open valve to remove pressure in the cylinder. Then

puncture, drill, crush, or otherwise destroy empty cylinder and dispose

of in a facility permitted for nonhazardous waste. Return all other

containers to supplier.

Refrigeration Application: Subject to "no venting" regulations of Sections 608 and 609 of the

Clean Air Act during the service or disposal of equipment.

### **SECTION 14 - TRANSPORT INFORMATION**

UN number (DOT, TDG, IMDG, UN 3159

IATA):

UN proper shipping name (DOT,

1,1,1,2-tetrafluoroethane or Refrigerant gas R 134a

TDG, IMDG, IATA):

Hazard class (DOT, TDG, IMDG, 2.2

IATA):

Packing group (DOT, TDG,

IMDG, IATA):

None

Hazardous substance (RQ): None



Environmental hazards (e.g.,

Marine pollutant):

Not a Marine Pollutant

Placard/label:

Non-flammable gas

Transport in bulk (according to Annex II of MARPOL 73/78 and the IBC Code):

Not available

Special precautions which a user needs to be aware of, or needs to comply with, in connection with transport or conveyance either within or outside their premises:

Consult applicable regulations (e.g., DOT, TDG, IATA, IMDG) for special precautions applicable to transport outside of user's premises. Within user's premises transport in upright, closed, and secured containers.

#### **SECTION 15 - REGULATORY INFORMATION**

**USA Classification** This material is classified as hazardous under OSHA regulations (29

CFR 1910.1200).

TSCA (Toxic Substances Control Act) Regulations, 40 CFR 710:

This product is listed on the TSCA Chemical Substances Inventory.

CERCLA and SARA

Regulations:

40 CFR 372: This product does not contain any chemicals subject to

reporting requirements of SARA Section 313.

40 CFR 355: This product does not contain any "extremely hazardous

chemical" subject to the requirements of SARA Section 312.

40 CFR 370: Hazardous properties as defined under the Hazard

Communication Standard (29 CFR 1910.1200):

Immediate (acute) health hazard, Sudden release of pressure.

Actions may be necessary under SARA Sections 311 and 312.

Consult regulations for applicability.

Ozone Protection and 40 CFR

82:

This product does not contain nor is it manufactured with ozone

depleting substances.

Other regulations/legislation: Subject to "no venting" regulations of Sections 608 and 609 of the

Clean Air Act during the service or disposal of equipment.

Canadian Classification: This product has been classified according to the hazard criteria of

the Controlled Product Regulations (CPR) and the SDS contains all

the information required by the CPR.

Controlled Products Regulations

(WHMIS Classification):

Class A: Compressed Gas

CEPA/Canadian Domestic Substances List (DSL):

The substance in this product is on the Canadian Domestic

Substance List (CEPA DSL).





Other regulations/legislation: This product contains the following substances present on the CEPA

2014 list of greenhouse gases subject to mandatory reporting:

1,1,1,2-tetrafluoroethane.

#### SECTION 16 - OTHER INFORMATION, INCLUDING DATE OF PREPARATION OR LAST REVISION

Prepared by: James W. Hathcock, Mexichem Fluor Inc.

Telephone number of preparer: 225-642-0094

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Version: 4

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